

PACE-IIT & MEDICAL

MUMBAI / DELHI-NCR / PUNE / NASHIK / AKOLA / GOA / JALGOAN / BOKARO / AMRAVATI / DUBAI / DHULE

IIT – JEE: 2026

TW TEST (3 YRS.)

DATE: 30/07/23

TOPIC: VECTORS
CALCULUS IN PHYSICS

ANSWER KEY

- | | | | | |
|---------|---------|---------|---------|---------|
| 1. (C) | 2. (D) | 3. (A) | 4. (A) | 5. (C) |
| 6. (C) | 7. (A) | 8. (A) | 9. (C) | 10. (A) |
| 11. (B) | 12. (C) | 13. (B) | 14. (A) | 15. (A) |
| 16. (B) | 17. (C) | 18. (B) | 19. (A) | 20. (D) |

SOLUTIONS

1. (C)

$$\frac{d(x^n)}{dx} = nx^{n-1}$$

2. (D)

$$\frac{d(\cos x - x \ln x + x)}{dx} = -\sin x - \left(\ln x + x \cdot \frac{1}{x} \right) + 1$$

3. (A)

$$\frac{d\left(\frac{1+x}{1-x}\right)}{dx} = \frac{1(1-x) - (-1)(1+x)}{(1-x)^2}$$

4. (A)

$$a = \frac{dv}{dt} = 6t + 2 = 14 \text{ at } t = 2$$

5. (C)

$$\frac{dy}{dx} = 1 - \frac{1}{x^2}; \frac{d^2y}{dx^2} = +\frac{2}{x^3}$$

$$\text{For minimum, } \frac{dy}{dx} = 0 \Rightarrow x = \pm 1$$

$$\text{At } x = 1, \frac{d^2y}{dx^2} = \frac{+2}{1^3} > 0$$

So, y is min at $x = 1$.

$$y_{\min} = 1 + \frac{1}{1} = 2$$

6. (C)

$$\int \left(\cos x - \frac{1}{x} + 1 \right) dx = \sin x - \ln x + x + C$$

7. (A)

Let $Z = \sec x \Rightarrow \frac{dz}{dx} = \sec x + \tan x$
 $\Rightarrow dz = \sec x \tan x dx$
 So, integral is $\int \sec x \cdot \sec x \tan x dx$
 $\int z dx = \frac{z^2}{2} + C = \frac{\sec^2 x}{2} + C$

8. (A)

$$\int \frac{3x^2 + 2}{x^3 + 2x + 8} dx$$

Let, $P = x^3 + 2x + 8$
 $\frac{dP}{dx} = 3x^2 + 2$
 $\Rightarrow \frac{3x^2 + 2}{x^3 + 2x + 8} dx = \int \frac{dP}{P} = \ln P + C$
 $= \ln(x^3 + 2x + 8) + C$

9. (C)

$$y = \ln P^2 = 2 \ln P$$

$$\Rightarrow \frac{dy}{dP} = \frac{2}{P}$$

$$x = \frac{P^3}{3} \Rightarrow \frac{dx}{dP} = P^2$$

$$\frac{dy}{dx} = \frac{dy}{dP} \times \frac{dP}{dx} = \frac{2}{P} \times \frac{1}{P^2} = \frac{2}{P^3}$$

10. (A)

$$v = 3t^2 - 2t + 5$$

$$\Rightarrow dS = 3 \int t^2 dt - 2 \int t dt + 5 \int dt$$

$$[S]_0^s = 3 \left[\frac{t^3}{3} \right]_2^3 - 2 \left[\frac{t^2}{2} \right]_2^3 + 5[t]_2^3$$

$$S = (27 - 8) - (9 - 4) + 5(3 - 2)$$

$$S = 19 - 5 + 5 = 19$$

11. (B)

$$R = \sqrt{3^2 + 4^2 + 2 * 3 * 4 \cos 60} = \sqrt{37}$$

12. (C)

$$|\overline{AB}| = \sqrt{6^2 + 2^2} = \sqrt{40}$$

13. (B)

14. (A)

$$\hat{A} = \frac{\vec{A}}{|\vec{A}|}$$

15. (A)

$$\left| \frac{\hat{i}}{\sqrt{2}} + \frac{\hat{j}}{\sqrt{2}} \right| = \sqrt{\frac{1}{2} + \frac{1}{2}} = 1$$

16. (B)

$$\vec{P} \cdot \vec{Q} = a^2 - 2a - 3 = 0$$
$$a = 3$$

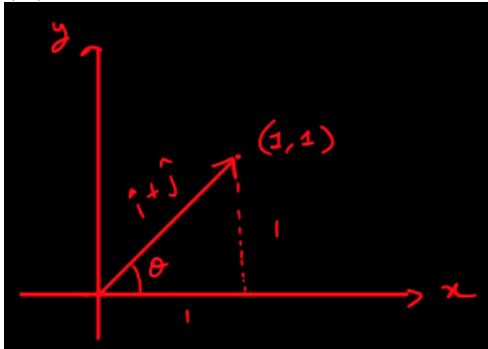
17. (C)

$$\hat{A} = \frac{\vec{A}}{A} = \frac{2\hat{i} - \hat{j} + \hat{k}}{\sqrt{2^2 + 1^2 + 1^2}} = \frac{2\hat{i} - \hat{j} + \hat{k}}{\sqrt{6}}$$

18. (B)

Both are opposite. Hence angle between them π .

19. (A)



$$\tan \theta = \frac{1}{1} = 1$$

$$\theta = 45^\circ$$

20. (D)

$$\vec{A} = \hat{i} - 2\hat{j} + \hat{k}$$

$$\vec{B} = 3\hat{i} + \hat{j} - 2\hat{k}$$

$$\vec{A} \times \vec{B} = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & -2 & 1 \\ 3 & 1 & -2 \end{vmatrix}$$

$$= \hat{i}(4-1) - \hat{j}(-2-3) + \hat{k}(1+6)$$

$$= 3\hat{i} + 5\hat{j} + 7\hat{k}$$

$$|\vec{A} \times \vec{B}| = \sqrt{3^2 + 5^2 + 7^2} = \sqrt{83}$$

$$\text{So, } \frac{\vec{A} \times \vec{B}}{|\vec{A} \times \vec{B}|} = \frac{3\hat{i} + 5\hat{j} + 7\hat{k}}{\sqrt{83}}$$

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DATE: 30/07/23

TOPIC: MOLE CONCEPT

Answer Key

- | | | | | |
|---------|---------|---------|---------|---------|
| 21. (A) | 22. (A) | 23. (B) | 24. (C) | 25. (D) |
| 26. (B) | 27. (C) | 28. (C) | 29. (B) | 30. (B) |
| 31. (B) | 32. (B) | 33. (B) | 34. (D) | 35. (C) |
| 36. (A) | 37. (D) | 38. (B) | 39. (B) | 40. (D) |

Solution

21. (A)

Atomic mass in gram = Mass of 6.022×10^{23} atoms

$$= \frac{1.15 \times 6.022 \times 10^{23}}{3.011 \times 10^{22}} = 23$$

22. (A)

$$\%N = \frac{\text{Mass of nitrogen}}{\text{Molecular mass}} \times 100\%$$

$$20 = \frac{14}{m} \times 100\%$$

$$\therefore m = 70$$

23. (B)

$$\text{Initiated moles} = \frac{0.098}{98} = 10^{-3}$$

$$\text{Removed moles of H}_2\text{SO}_4 = \frac{3.01 \times 10^{20}}{6.02 \times 10^{23}} = 0.5 \times 10^{-3}$$

$$\text{Remaining moles of H}_2\text{SO}_4 = 10^{-3} - 0.5 \times 10^{-3} = 0.5 \times 10^{-3}$$

24. (C)

$$\text{Mass of NaCl in 10 g salt} = \frac{10 \times 95}{100} = 9.5 \text{ g}$$

$$\therefore \text{Number of molecules of NaCl} = \frac{9.5}{58.5} \times 6.023 \times 10^{23} = 9.78 \times 10^{22} \approx 10^{23}$$

25. (D)
 $\text{Na}_2\text{SO}_4 \cdot n\text{H}_2\text{O}$
 Molar mass = $(142 + 18n)$
 \therefore Mass of water = $\frac{12.6}{26.8} \times (142 + 18n)$
 $18n = \frac{12.6}{26.8} \times (142 + 18n)$
 $\therefore n = 7$

26. (B)

Element	%	Atomic ratio	Simplest ratio
X	50	$\frac{50}{10} = 5$	$\frac{5}{2.5} = 2$
Y	50	$\frac{50}{20} = 2.5$	$\frac{2.5}{2.5} = 1$

Formula = X_2Y

27. (C)
 $4\text{PH}_3(\text{s}) \rightarrow \text{P}_4(\text{g}) + 6\text{H}_2(\text{g})$
 $4 \text{ mL} \rightarrow 1 \text{ mL} \quad 6 \text{ mL}$
 $100 \text{ mL} \rightarrow \frac{100}{4} \quad \frac{6}{4} \times 100$
 $100 \text{ mL} \rightarrow 25 \text{ mL} \quad 150 \text{ mL}$
 Volume increase by 75 mL

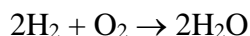
28. (C)
 Let x L liquid contain 4 mole in it
 Number of moles = $\frac{\text{Mass}}{\text{Molar mass}}$
 $4 = \frac{x \times 100 \times 1.4}{280}$
 $\therefore x = 0.8\text{L}$

29. (B)
 No. of moles of $\text{Na}^+ = 3 \times$ No. of moles of Na_3PO_4
 $= 3 \times \frac{0.4 \times 20}{1000} = 0.024$

30. (B)
 No. of moles of sucrose = $\frac{25.6 \text{ g}}{342.3 \text{ g}} = 0.075$
 No. of moles of hydrogen atom = 0.075×22
 No. of atoms of H = $0.075 \times 22 \times 6.023 \times 10^{23} = 9.9 \times 10^{23}$

31. (B)

$$n_{\text{H}_2} = \frac{10}{2} = 5, \quad n_{\text{O}_2} = \frac{64}{32} = 2$$



32. (B)

$$\text{No. of moles of H}_2\text{O} = \frac{0.72}{18} = 0.04 \quad n_{\text{H}} = 2 \times n_{\text{H}_2\text{O}} = 2 \times 0.04 = 0.08$$

$$\text{No. of moles of CO}_2 = \frac{3.08}{44} = 0.07 \quad n_{\text{C}} = n_{\text{CO}_2} = 0.07$$

$$\text{C} : \text{H} = 7 : 8$$

33. (B)

$$\text{Mass} = 18 \text{ g}$$

$$\text{No. of molecules in 18 g H}_2\text{O} = 6.023 \times 10^{23}$$

$$\text{No. of electrons in 18 g water} = 6.023 \times 10^{23} \times 10 = 6.023 \times 10^{24}$$

34. (D)

$$\text{Charge on 1 mol Al}^{3+} = 3e N_{\text{A}} \text{ coulomb}$$

35. (C)

$$\text{Volume of 1 drop} = \frac{2}{35} \text{ mL}$$

$$\text{Number of moles in 1 drop} = \frac{2 \times 1.2}{35 \times 70} = \frac{1.2}{(35)^2}$$

$$\text{Number of molecules in one drop} = \frac{1.2}{(35)^2} \times N_{\text{A}}$$

36. (A)

$$0.2 = \frac{n_{\text{glucose}}}{n_{\text{glucose}} + n_{\text{H}_2\text{O}}}$$

$$\text{Let } n_{\text{glucose}} = x$$

$$\Rightarrow n_{\text{H}_2\text{O}} = 4x$$

$$\text{Molality} = \frac{n_{\text{glucose}}}{W_{\text{H}_2\text{O}} (\text{kg})} = \frac{x}{4x \times \frac{18}{1000}} = \frac{250}{18} = 13.8$$

37. (D)

For minimum MW, number of oxygen atoms per molecule = 1

$$\text{So } 3.2 = \frac{16 \times 1}{MW_{\text{min}}} \times 100$$

$$\Rightarrow MW_{\text{min}} = \frac{1600}{3.2} = 500$$

38. (B)
 $x+1+2 \times 2 = y \times 1.5$
 $\Rightarrow x+4=1.5y$
Since final molarity is average of both molarities, the volume of both solution has to be equal
 $\Rightarrow x=2$
 $\Rightarrow y=4$

39. (B)
 $n_{\text{Cl}^-} = 2 \times 1 \times 2 + 2 \times 1 \times 1 + 1 \times 1 \times 3$
 $= 4 + 2 + 3 = 9$
 $[\text{Cl}^-] = \frac{9}{5} = 1.8 \text{ M}$

40. (D)
 $n_{\text{OH}^-} = V \times \frac{1}{10} \times 2$
 $n_{\text{H}^+} = 0.5 \times 2 \times 2$
 $V \times \frac{1}{10} \times 2 = 0.5 \times 2 \times 2$
 $V = 10 \text{ L}$

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TOPIC: TRIGONOMETRY-I

ANSWER KEY

41. (C)	42. (B)	43. (B)	44. (C)	45. (C)
46. (D)	47. (A)	48. (D)	49. (B)	50. (A)
51. (A)	52. (B)	53. (C)	54. (C)	55. (B)
56. (C)	57. (D)	58. (B)	59. (B)	60. (A)

SOLUTIONS

41. (C)

$$\begin{aligned} & \cos 15^\circ - \sin 15^\circ \\ &= \frac{\sqrt{3}+1}{2\sqrt{2}} - \left(\frac{\sqrt{3}-1}{2\sqrt{2}} \right) = \frac{1}{\sqrt{2}} \end{aligned}$$

42. (B)

$$\text{Since } \operatorname{cosec}^2 \theta = 1 + \cot^2 \theta = 1 + \frac{9}{16} = \frac{25}{16} \quad \left(\because \tan \theta = -\frac{4}{3} \right)$$

$$\sin^2 \theta = \frac{1}{\operatorname{cosec}^2 \theta} = \frac{16}{25} \Rightarrow \sin \theta = \pm \frac{4}{5},$$

Both the values are acceptable, since $\tan \theta = -\frac{4}{3}$

i.e., θ lies in 2nd or 4th quadrant.

43. (B)

$$\begin{aligned} \therefore \cos x + \cos(120^\circ - x) + \cos(120^\circ + x) &= 2 \cdot \cos 120^\circ \cdot \cos x + \cos x \\ &= 2 \cdot \cos(90^\circ + 30^\circ) \cdot \cos x + \cos x = 2(-\sin 30^\circ) \cdot \cos x + \cos x \\ &= \left(2 - \frac{1}{2} \right) \cdot \cos x + \cos x = -\cos x + \cos x = 0 \end{aligned}$$

44. (C)

$$\cos \frac{\pi}{6} = \frac{\sqrt{3}}{2}$$

$$\cos \frac{5\pi}{6} = -\frac{\sqrt{3}}{2}$$

$$\cos 7\pi/6 = -\frac{\sqrt{3}}{2}$$

$$\cos 19\pi/6 = -\sqrt{3}/2$$

45. (C)

$$\sin \theta = -\frac{1}{\sqrt{2}} \text{ and } \tan \theta = 1$$

$$\Rightarrow \sin \theta = \sin 225^\circ \Rightarrow \theta = 225^\circ$$

Since $\sin \theta$ is -ve and $\tan \theta$ is +ve in third quadrant.

46. (D)

$$3 \tan A + 4 = 0 \Rightarrow \tan A = -\frac{4}{3}$$

$$\Rightarrow \sin A = \pm \frac{\tan A}{\sqrt{1 + \tan^2 A}} = \pm \frac{-4/3}{\sqrt{1 + 16/9}} = \frac{4}{5} \quad (\because A \text{ is in } 2^{\text{nd}} \text{ quadrant})$$

$$\text{and } \cos A = -\frac{3}{5}. \text{ Thus, } 2 \cot A - 5 \cos A + \sin A = 2\left(-\frac{3}{4}\right) - 5\left(-\frac{3}{5}\right) + \frac{4}{5} = \frac{23}{10}.$$

47. (A)

$$\tan(90^\circ - \theta) = \cot \theta, \cot(90^\circ - \theta) = \tan \theta.$$

48. (D)

$$\text{We have } \sin A = \frac{4}{5} \text{ and } \cos B = -\frac{12}{13}$$

$$\text{Now, } \cos(A+B) = \cos A \cos B - \sin A \sin B$$

$$= \sqrt{1 - \frac{16}{25}} \left(-\frac{12}{13}\right) - \frac{4}{5} \sqrt{1 - \frac{144}{169}}$$

$$= -\frac{3}{5} \times \frac{12}{13} - \frac{4}{5} \left(-\frac{5}{13}\right) = -\frac{16}{65}$$

(Since A lies in first quadrant and B lies in third quadrant).

49. (B)

$$\tan 20^\circ + 2 \tan 50^\circ - \tan 70^\circ$$

$$= \frac{\sin 20^\circ}{\cos 20^\circ} - \frac{\sin 70^\circ}{\cos 70^\circ} + 2 \tan 50^\circ$$

$$= \frac{\sin 20^\circ \cos 70^\circ - \cos 20^\circ \sin 70^\circ}{\cos 20^\circ \cos 70^\circ} + 2 \tan 50^\circ$$

$$= \frac{\sin(20^\circ - 70^\circ)}{\cos 20^\circ \cos 70^\circ} + 2 \tan 50^\circ$$

$$= \frac{1}{2} [\cos(70^\circ + 20^\circ) + \cos(70^\circ - 20^\circ)]$$

$$= \frac{2 \sin(-50^\circ)}{\cos 90^\circ + \cos 50^\circ} + 2 \tan 50^\circ$$

$$= \frac{-2 \sin 50^\circ}{0 + \cos 50^\circ} + 2 \tan 50^\circ$$

$$= -2 \tan 50^\circ + 2 \tan 50^\circ = 0.$$

50. (A)
 $\tan 75^\circ - \cot 75^\circ = \cot 15^\circ - \cot 75^\circ$
 $= (2 + \sqrt{3}) - (2 - \sqrt{3}) = 2\sqrt{3}.$
51. (A)
 $\sin 330^\circ = -1/2$
 $\tan 225^\circ = +1$
 $\cos 210^\circ = -\sqrt{3}/2$
 Ans. (A)
52. (B)
 $3A = 2A + A$
 $\tan 3A = \tan(2A + A)$
 $= \frac{\tan 2A + \tan A}{1 - \tan A \tan 2A}$
 $\therefore \tan 3A - \tan 2A \tan A \tan 2A = \tan 2A + \tan A$
 $\therefore \tan 3A \tan 2A \tan A = \tan 3A - \tan 2A - \tan A$
53. (C)
 $(1 + \tan 11^\circ)(1 + \tan 22^\circ)(1 + \tan 34^\circ)(1 + \tan 23^\circ)$
 $= (1 + \tan 11^\circ)(1 + \tan 34^\circ)(1 + \tan 22^\circ)(1 + \tan 23^\circ)$
 $= 2(2) = 4$
54. (C)
 Substitute the corresponding values.
55. (B)
 $3[\cos^4 \alpha + \sin^4 \alpha] - 2[\cos^6 \alpha + \sin^6 \alpha] = 3[1 - 2\sin^2 \alpha \cos^2 \alpha] - 2[1 - 3\sin^2 \alpha \cos^2 \alpha] = 1$
56. (C)
 If $(A + B) = \frac{\pi}{2}$ then $(\sin^2 A + \sin^2 B) = 1$
57. (D)

$$\frac{\cos(\theta - 90^\circ) \sec \theta \tan(180^\circ - \theta)}{\sec(360^\circ - \theta) \sin(\theta - 180^\circ) \tan(360^\circ - \theta)}$$

$$= \frac{\sin \theta (\sec \theta) (-\tan \theta)}{\sec \theta (-\sin \theta) (-\tan \theta)} = -1$$
58. (B)
 $1358^\circ = 8(180^\circ) + 82^\circ, 3608^\circ = 20(180^\circ) + 8$

59. (B)

If $(A + B) = 90^\circ$ then $(\sin^2 A + \sin^2 B) = 1$

60. (A)